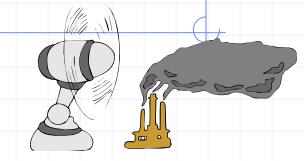
Fluids







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Basic Properties

Pressure

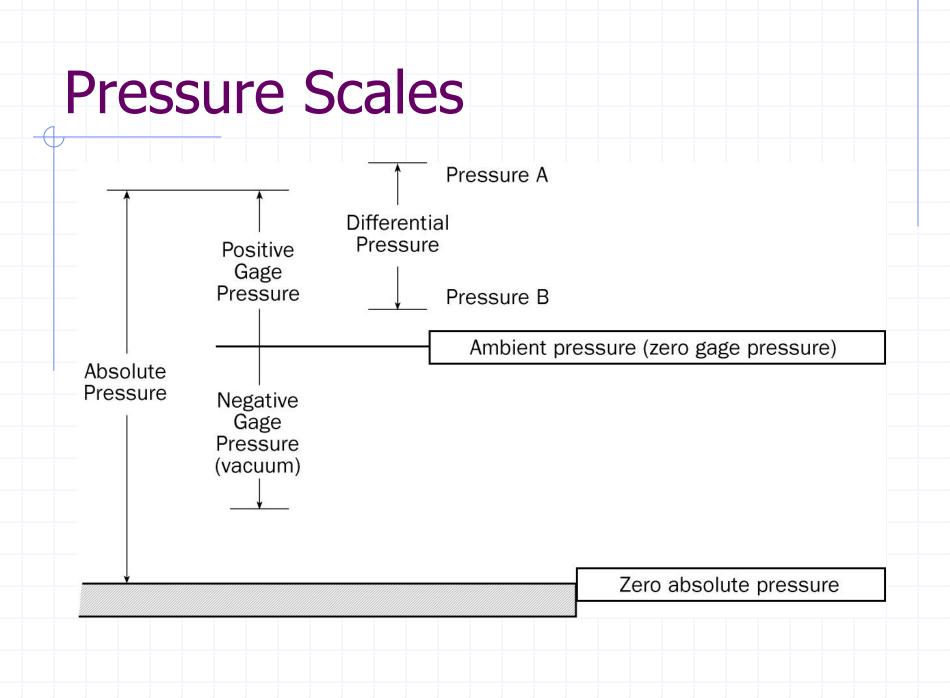
 Unit – Pascal (N/m²)

 Density

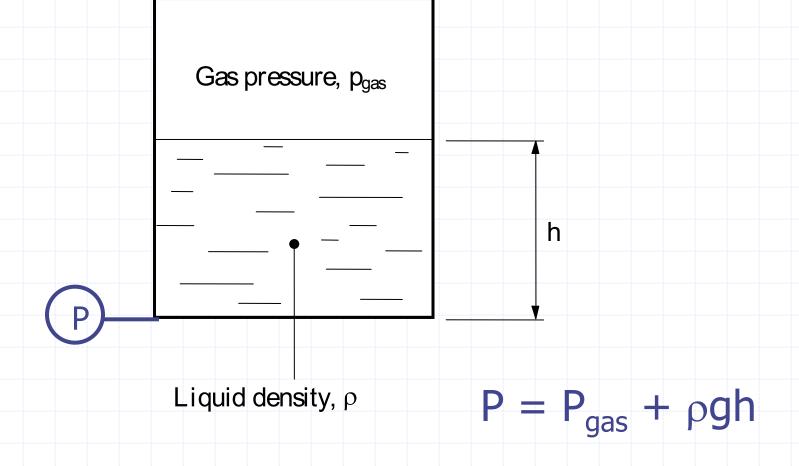
 Unit – kg/m³

 Viscosity

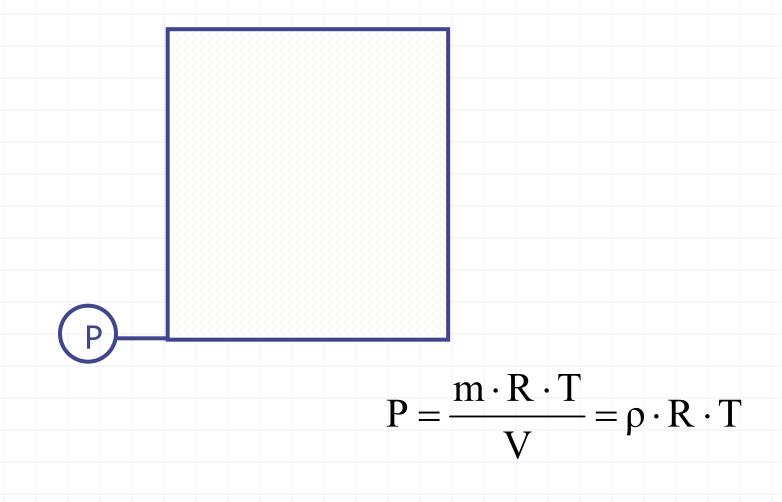
 Unit – poise (Pa's)



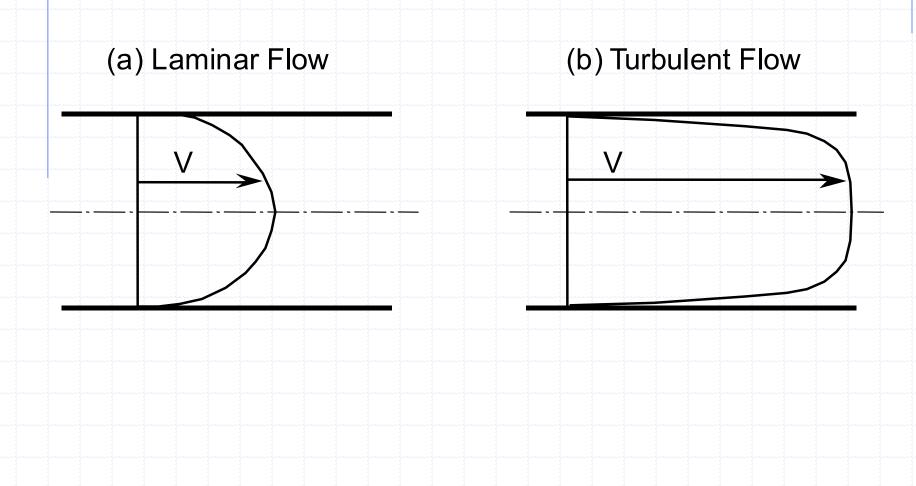
Factors Affecting Pressure (Liquids)



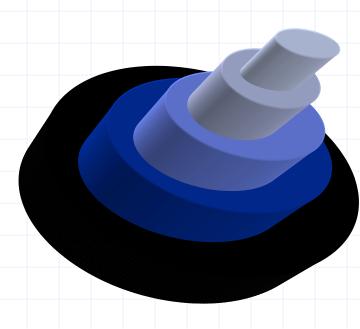
Factors Affecting Pressure (gas)



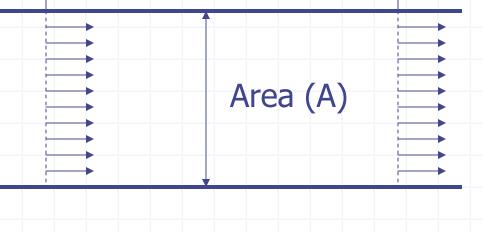
Laminar and Turbulent Flow

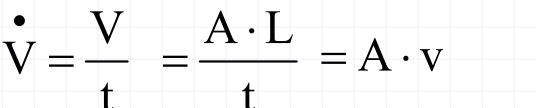


Laminar Flow



Volumetric Flow Rate



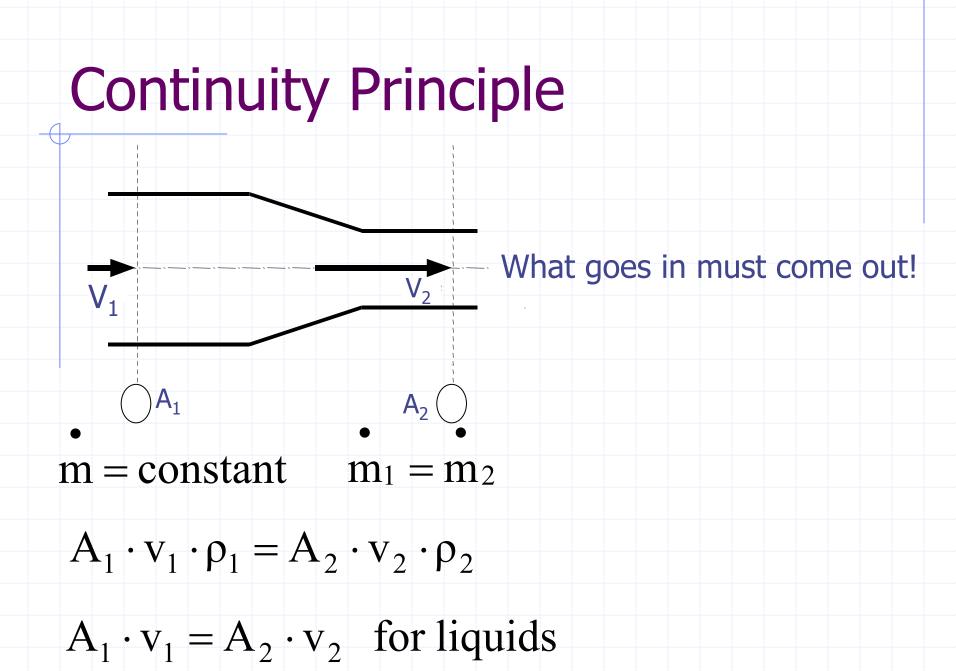


Mass Flow

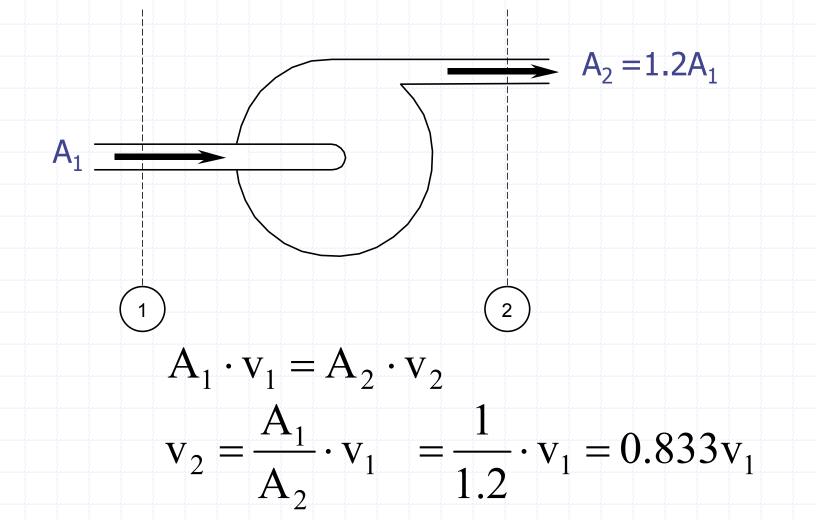
$\rho = \frac{m}{V} \longrightarrow m = V \cdot \rho$

$\mathbf{e} \quad \mathbf{e} \\ \mathbf{m} = \mathbf{V} \cdot \mathbf{\rho} = \mathbf{A} \cdot \mathbf{v} \cdot \mathbf{\rho}$





What is the relationship between v_1 and v_2 ?



Temperature and Pressure

What happens to volume flow rates?
 Temperature increase

 Liquid
 Gas

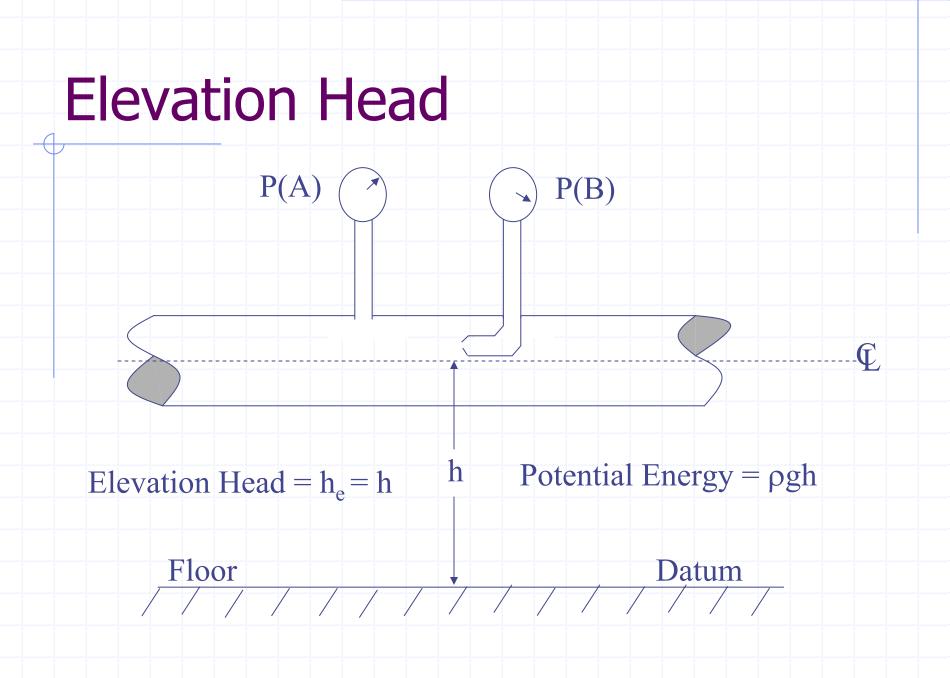
 Pressure Increase

 Liquid
 Gas

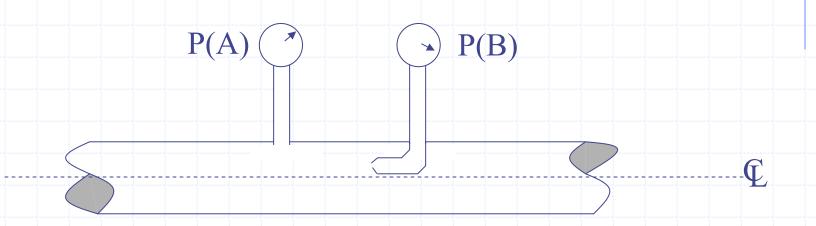
Energy In a Flowing Fluid

Kinetic energy of moving fluid
 Potential energy due to elevation

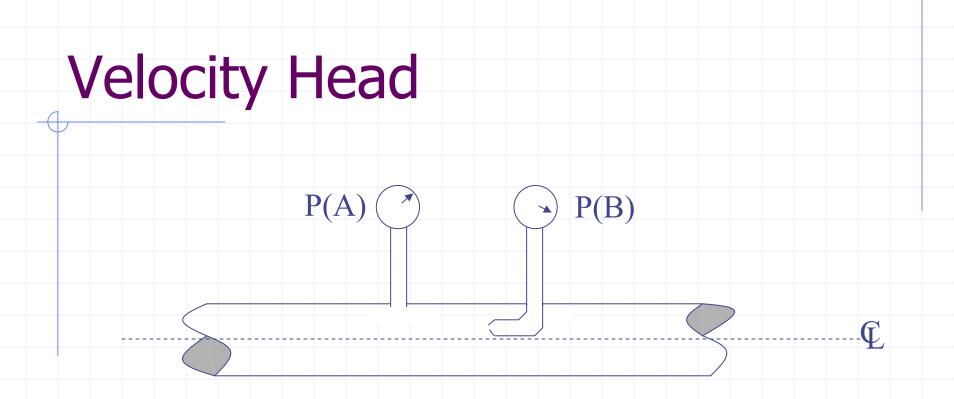
- Pressure energy
 - A measure of the work done on the fluid to push it into the pipe
 - Or a measure of work done by the liquid when it comes out of the pipe
- Internal energy
 - Heat, intermolecular forces, intra-molecular forces







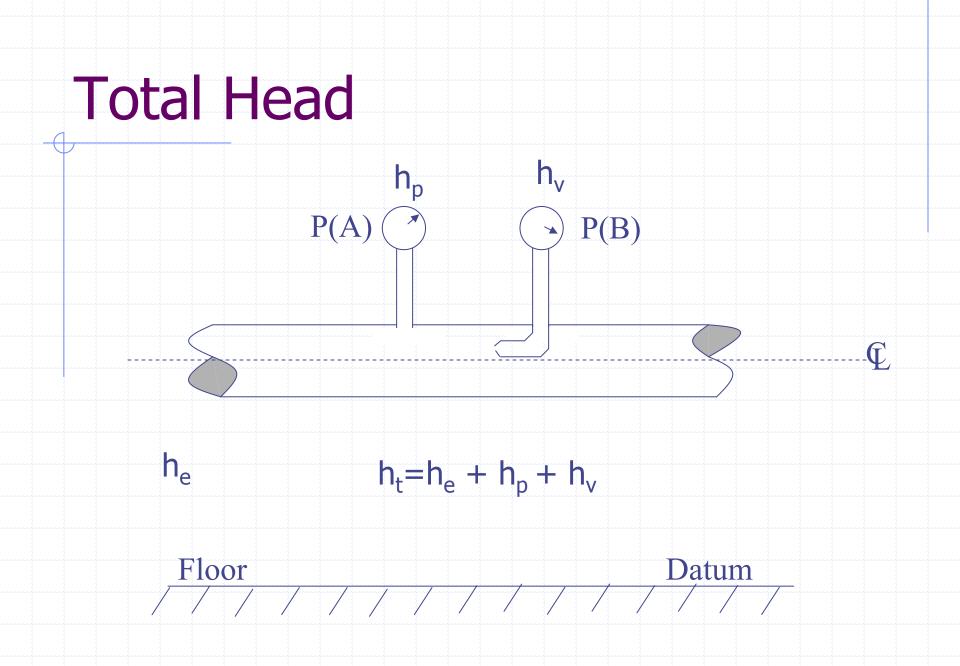
P(A) static pressure of system Column of water supported by this pressure Pressure Head $h_p = \frac{p}{\rho g}$



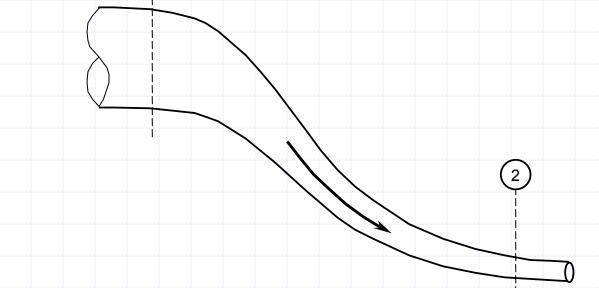
P(B) is the static pressure plus the pressure of stopping the fluid Change the kinetic energy of the fluid to potential energy

$$n_v = \frac{v}{2g}$$

2

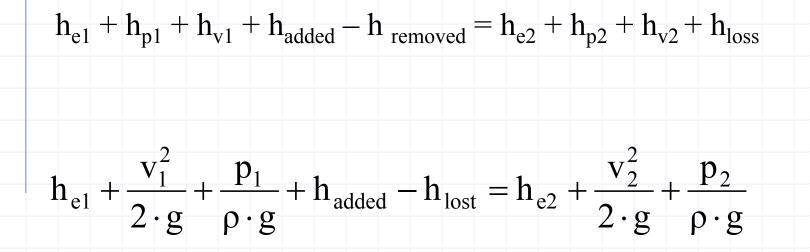


Conservation of Energy



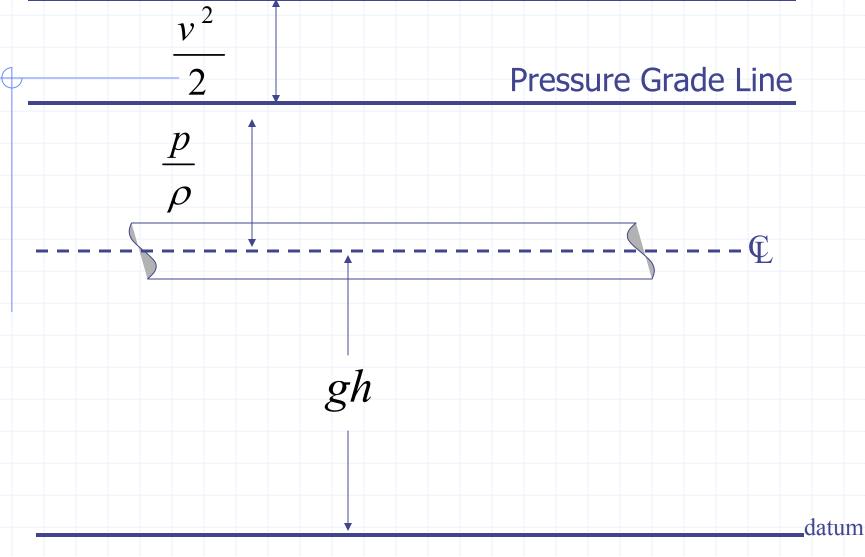
$$h_{e1} + h_{p1} + h_{v1} + h_{added} - h_{removed} = h_{e2} + h_{p2} + h_{v2} + h_{loss}$$

Bernoulli's Equation



Energy is conserved in a flowing fluid!

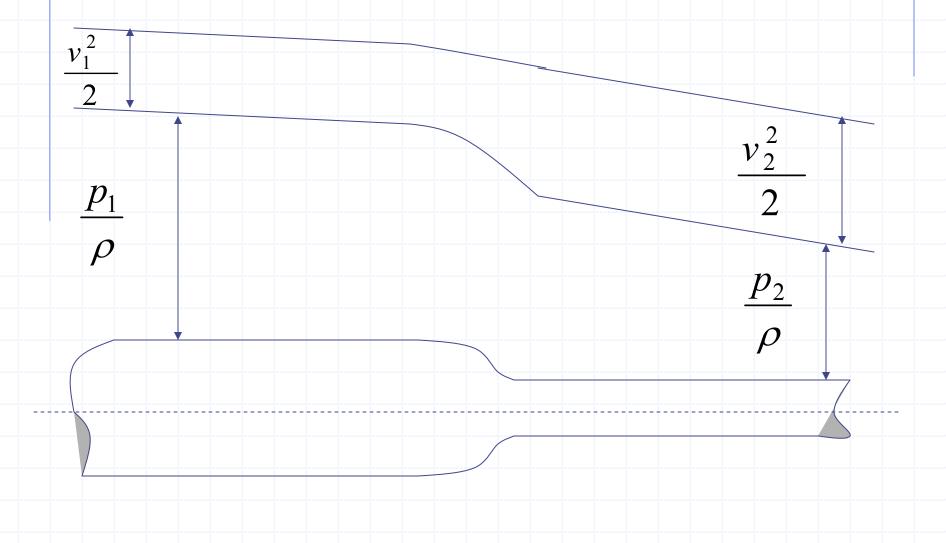


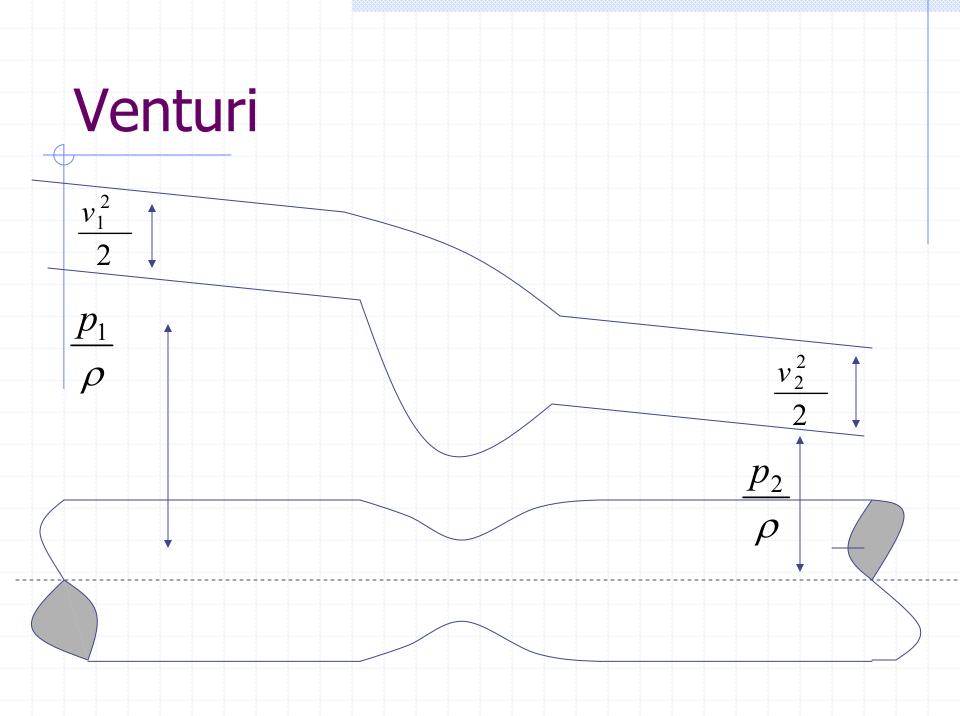




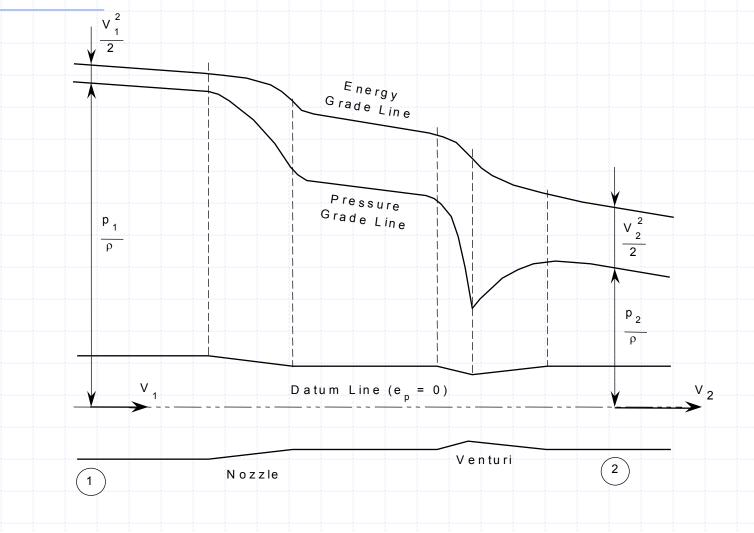
Energy v_{1}^{2} v_2^2 2 Pressure 2 p_1 p_2 Potential ρ ρ gh_1 gh_2 datum Energy Loss Due to Flow

Changes at a reduction

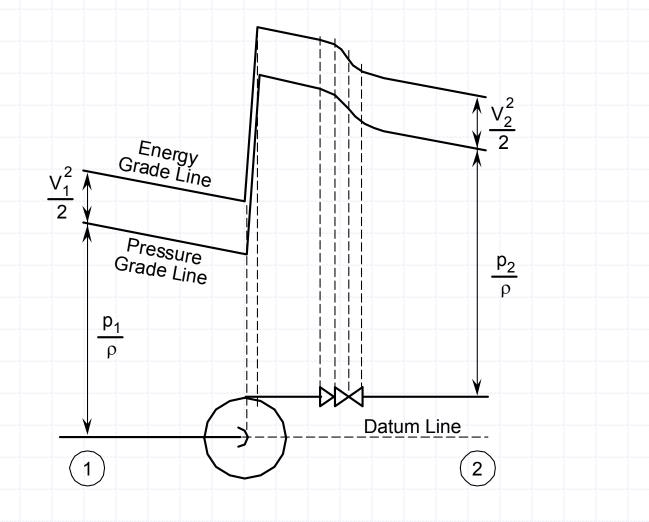




Reduction and Venturi

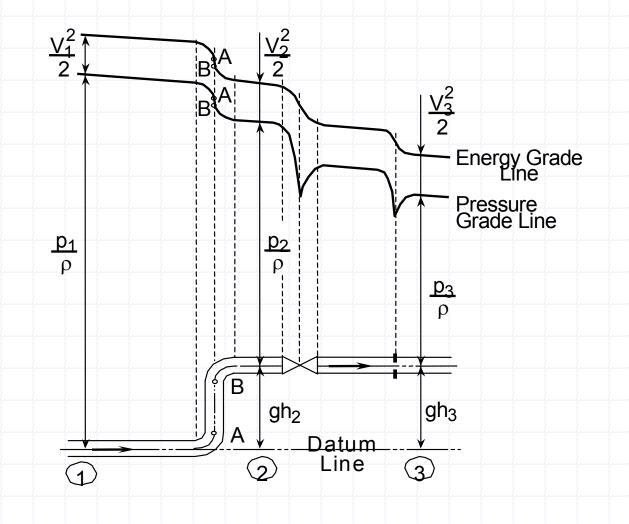


Pump and Discharge Valves

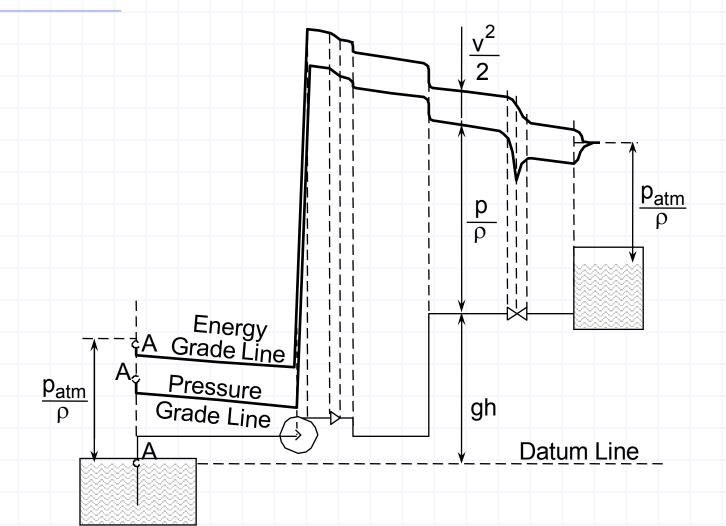


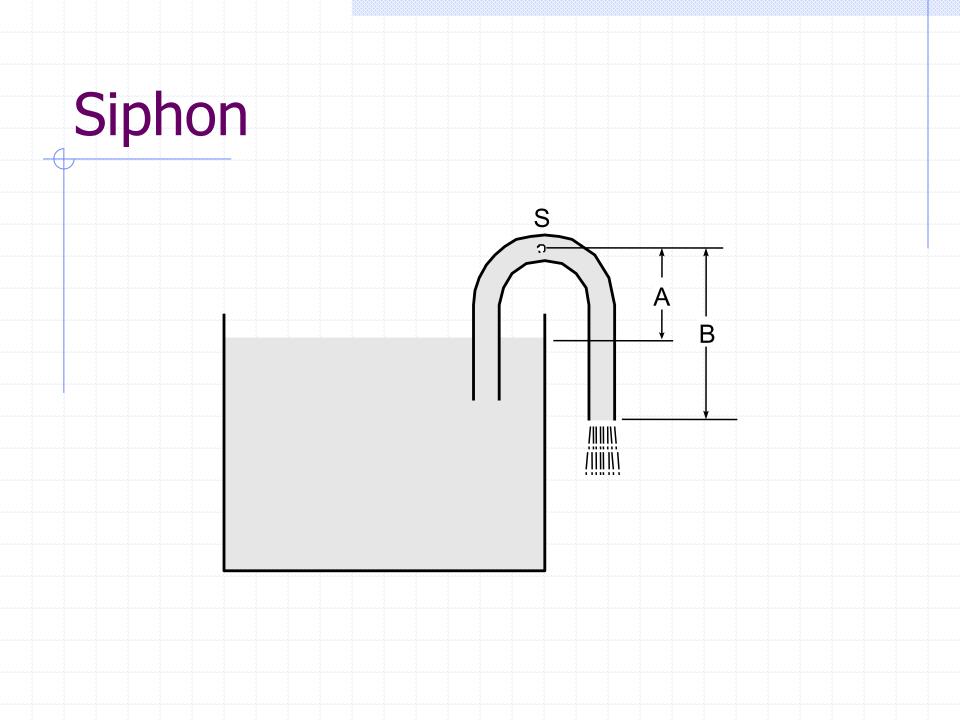
Elevation

4

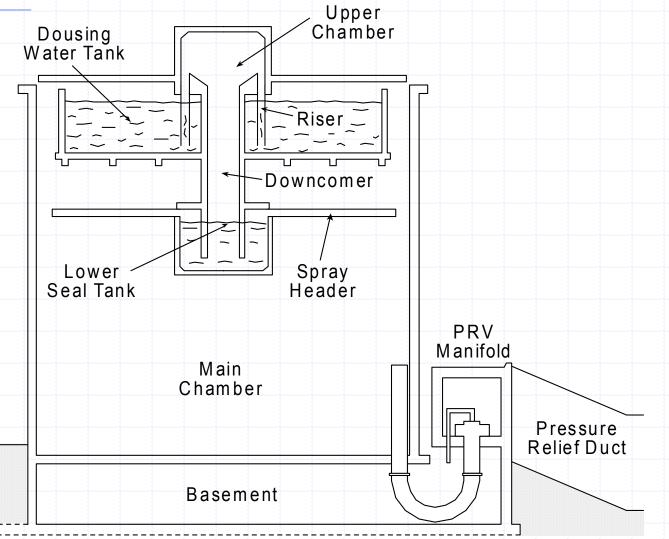


Simple System



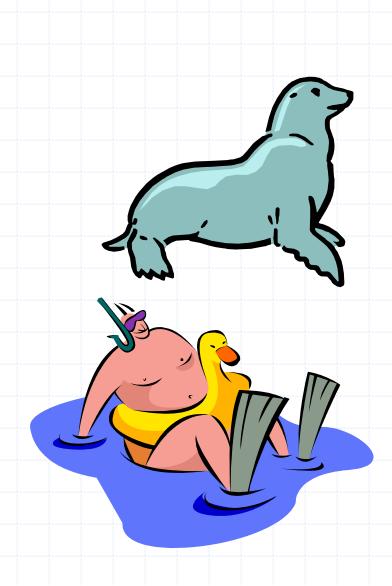


Dousing



Other Topics





Two Phase Flow

Flow of liquid and gas mixture

How Do We Get 2 Phase Flow?

Generate vapour Add heat to liquid Drop pressure Generate liquid Remove heat from gas Increase pressure Allow dissolved gas to come out of solution Entrain gas from a leak in

Forms of 2 phase Flow

Little gas bubbles in a liquid Large gas bubbles in a liquid A vapour film between a hot surface and cooling liquid Small liquid drops in gas flow Large liquid slugs in gas flow Stratified Flow

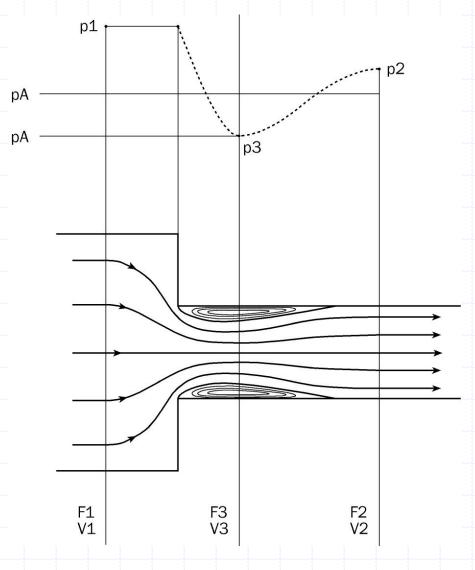
Cavitation

Pressure in a system drops below the saturation pressure Vapour bubbles form Bubbles flow to a place when the pressure is higher Bubbles collapse High speed jets of water appear

Cavitation 2



Cavitation



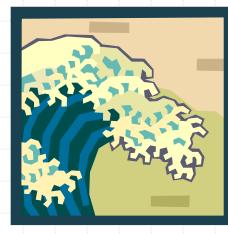
Potential Cavitation Sites

Suction channels of pumps
 Centrifugal pump impellers
 Sharp elbows and tees
 Down stream of partly open valves
 Sudden changes of flow area

Water Hammer

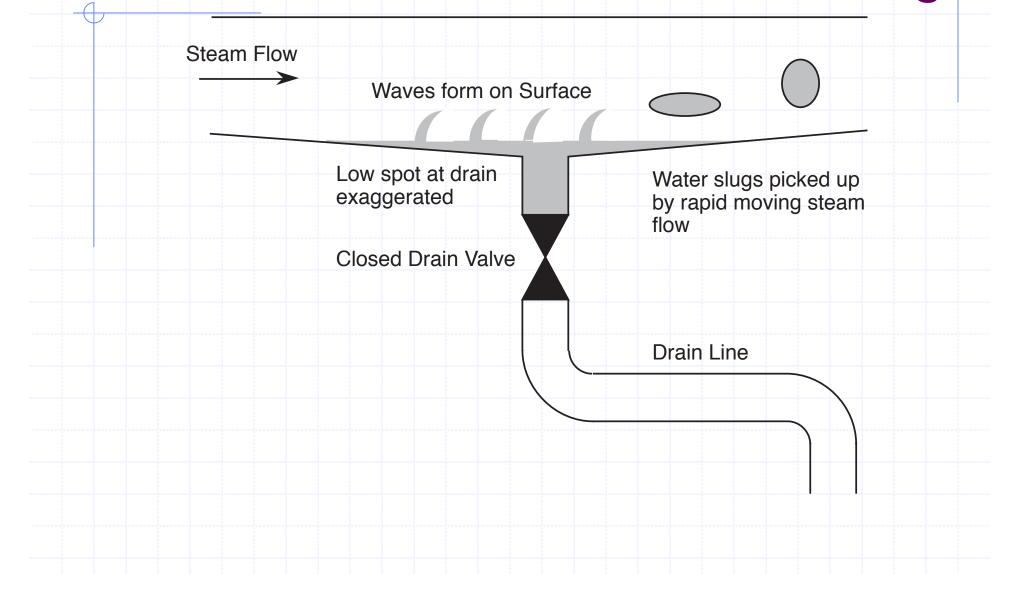
Quickly stopping water flowing in a pipe.
 Slugs of water thrown down a pipe by

steam flow hitting something solid



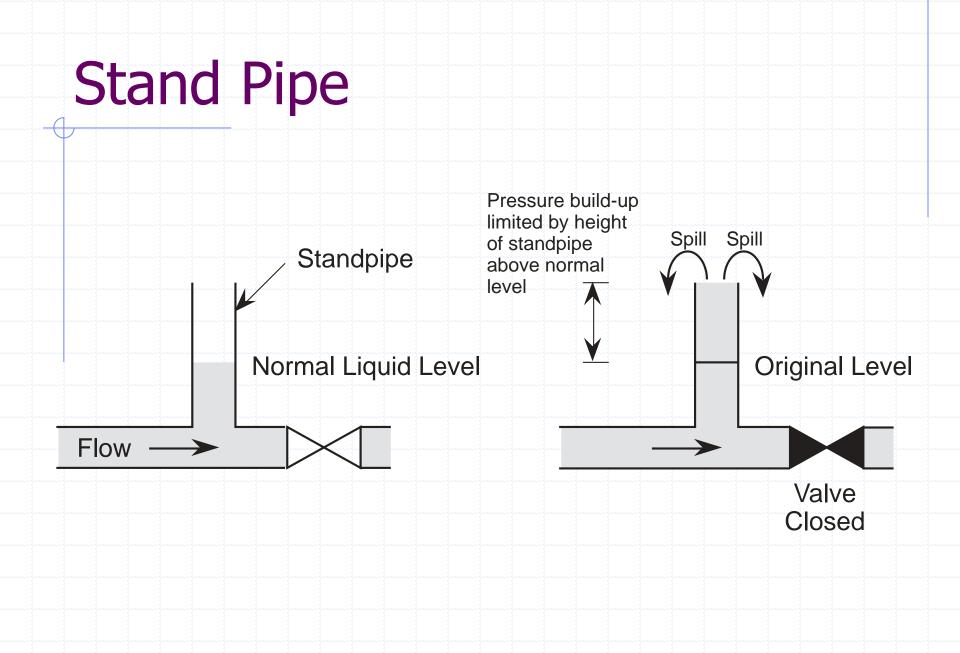


Water Hammer From Water Slugs



Minimizing Water Hammer

Orifice Plate
Water hammer arrestors
Standpipes



Steam Hammer

Steam bubbles collapse and water at high speed rushes in and then hits something solid

Good Operational Practices

Draining of a steam or gas system Venting and slow priming a liquid system Slow valve operation Starting centrifugal pumps with discharged valve closed Delays between pump starts or stops Apply cool water to Hx first

Solid Operation

Liquid System completely full of water or other liquid

- Non-compressible
- Susceptible to fast pressure transients
 - Water hammer
 - Velocity Changes

Flow Induced Vibrations

High turbulence
 Cavitation
 Pressure pulsation
 After effects of steam or water hammer

